



# Bottomdouble3

**Sebastian Brünink**



**Bottomdouble3:**

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 germany see also de in cantonese because the lack of

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 2 3 2 nacl 1 f 2 2 naf 1 cl 2 4 2 h 2 1 o 2 2 h 2 o 5 1 pb oh 2 2 hcl 2 h 2 o 1 pbcl 2 6 2 albr 3 3 k 2 so 4 6 kbr 1 al 2 so 4 3 7 1  
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 chemical equation  $2\text{Fe}_2\text{O}_3 + \text{C} + \text{Fe} + 3\text{CO}_2$  on towards the next step now step 3 once you are done with this check out if the number  
 of atoms for the other elements is also equal on both the sides

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 reactants and products mass molecular weight chemical equations formulas stoichiometric coefficient the stoichiometric  
 coefficient or stoichiometric number is the number of molecules that participate in the reaction

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 $\text{N}_2 + \text{F}_2 \rightarrow \text{NF}_3$  2  $\text{C}_6\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$  3  $\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$  4  $\text{Na}_2\text{SO}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{Na}_2\text{SO}_4 + \text{H}_2\text{O} + \text{SO}_2$  5  $\text{NF}_3 + \text{SNF}_2 \rightarrow \text{N}_2\text{O}_3$  using the equation  
 from problem 2 above answer the following

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