

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance
 (B) volume, molarity, pH, amount of substance
 (C) volume, molarity, pH, concentration of ions
 (D) amount of substance, pH, concentration of ions, concentration of substance

(A) $\text{pH} = 10.0$, $\text{pH} = 10.0$

(B) $10 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$, $1 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$, $10 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$
 (C) $1 \text{ mol H}_2\text{O}$, $10 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$

2. An acid-base reaction represents a proton transfer from a stronger acid to a weaker acid. Which of the following is not true? (correct)
- (A) The equilibrium constant is greater than 1.
 (B) The equilibrium constant is less than 1.
 (C) The reaction is exothermic.
 (D) The reaction is endothermic.



(B) $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{Cl}^-$, $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{Cl}^-$

3. An acid-base reaction involves the transfer of a proton from a stronger acid to a weaker acid. Which of the following is not true? (correct)
- (A) The equilibrium constant is greater than 1.
 (B) The equilibrium constant is less than 1.
 (C) The reaction is exothermic.
 (D) The reaction is endothermic.

(A) $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{Cl}^-$, $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{Cl}^-$

4. Which of the following is not a property of a strong acid? (correct)



(B) $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{Cl}^-$, $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{Cl}^-$

5. A solution of a weak acid and its conjugate base is prepared. The pH of the solution is 4.0. If the concentration of the weak acid is 0.10 M, what is the concentration of the conjugate base? (correct)



(B) 10^{-4} M , 10^{-4} M

6. An acid-base reaction involves the transfer of a proton from a stronger acid to a weaker acid. Which of the following is not true? (correct)
- (A) The equilibrium constant is greater than 1.
 (B) The equilibrium constant is less than 1.
 (C) The reaction is exothermic.
 (D) The reaction is endothermic.



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