

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- anode
- cathode
- anion
- cation
- electron movement flow
- direction of ion movement
- half-cell reduction equation
- Zn
- Write overall cell equation with phases. $\text{Zn(s)} + \text{Cu}^{2+}(\text{aq}) \rightarrow \text{Zn}^{2+}(\text{aq}) + \text{Cu(s)}$ (10 marks)

Voltaic Cells Study Guide

Kenneth L. Henold



Voltaic Cells Study Guide:

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