

# Chemistry: 2<sup>nd</sup> Semester Exam

Select the correct oxidation numbers for the following elements.

- Oxygen in  $O_2$ 
  - 2
  - +2
  - 1
  - +1
  - 0
- Sodium in  $NaCl$ 
  - 2
  - +2
  - 1
  - +1
  - 0
- Phosphorus in  $H_3PO_4$ 
  - 5
  - 3
  - 0
  - +3
  - +5
- H in  $H_2O$ 
  - 2
  - +2
  - 1
  - +1
  - 0
- O in  $H_2O$ 
  - 2
  - +2
  - 1
  - +1
  - 0
- In which substance does bromine have an oxidation number of +1?
  - $Br_2$
  - $HBr$
  - $HBrO$
  - $HBrO_2$
- Which species is the oxidizing agent in the following reaction?  $Cl_2(aq) + 2I^-(aq) \rightarrow I_2(aq) + 2Cl^-(aq)$ 
  - $Cl_2$
  - $2I^-$
  - $I_2$
  - $2Cl^-$
- Consider the following:  $2MnO_4^- + 5H_2SO_3 \rightarrow 2Mn^{2+} + 3H_2O + 5SO_4^{2-} + 4H^+$   
Which of the following elements does not experience a change in oxidation number?
  - S in  $H_2SO_3$
  - O in  $MnO_4^-$
  - Mn in  $MnO_4^-$
- In which of the following unbalanced equations does chromium undergo oxidation?
  - $Cr^{3+} \rightarrow Cr$
  - $Cr^{2+} \rightarrow Cr^{3+}$
  - $Cr^{3+} \rightarrow Cr_2O_7^{2-}$
  - $CrO_4^{2-} \rightarrow Cr_2O_7^{2-}$
- In a redox reaction, the species which loses electrons is \_\_\_\_\_.
  - reduced
  - oxidized
  - neutralized
- Consider the following:  $Cl_2 \rightarrow ClO_2^-$   
The balanced half reaction is:
  - $Cl_2 + 3H_2O \rightarrow 2ClO_2^- + 6H^+ + 5e^-$
  - $Cl_2 + 3H_2O \rightarrow 2ClO_2^- + 6H^+ + 4e^-$
  - $Cl_2 + 6H_2O + 2e^- \rightarrow 2ClO_2^- + 6H_2$
  - $Cl_2 + 6H_2O \rightarrow 2ClO_2^- + 12H^+ + 10e^-$

## Label the Acid, Base, Conjugate Acid, and Conjugate Base



- $HC_2H_3O_2$  serves as the
  - Acid
  - Base
  - Conjugate Acid
  - Conjugate Base
- $H_2O$  serves as the
  - Acid
  - Base
  - Conjugate Acid
  - Conjugate Base
- $H_3O^+$  serves as the
  - Acid
  - Base
  - Conjugate Acid
  - Conjugate Base
- $C_2H_3O_2^-$  serves as the
  - Acid
  - Base
  - Conjugate Acid
  - Conjugate Base

# 2nd Semester Exam Chemistry Form

**Arturo Cuomo**



**2nd Semester Exam Chemistry Form:**

The Top Books of the Year 2nd Semester Exam Chemistry Form The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous captivating novels captivating the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the captivating narratives that have charmed audiences this year. The Must-Read : Colleen Hoover "It Ends with Us" This heartfelt tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. 2nd Semester Exam Chemistry Form : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. 2nd Semester Exam Chemistry Form : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens crafts a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These popular novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Pape, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a exceptional and thrilling novel that will keep you wondering until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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