

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe the components and their function. (10 marks)

- a. Zn electrode
- b. ZnSO₄ solution
- c. Cu electrode
- d. CuSO₄ solution
- e. Salt bridge (connects the two solutions)
- f. External circuit (connects the two electrodes)
- g. Voltmeter (measures the potential difference)
- h. Zn
- i. Both Zn and Cu plates will be oxidized, but Cu will be reduced, as shown by the cell diagram with Zn as the anode and Cu as the cathode.

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Voltaic Cells Study Guide

B Lingard



Voltaic Cells Study Guide:

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