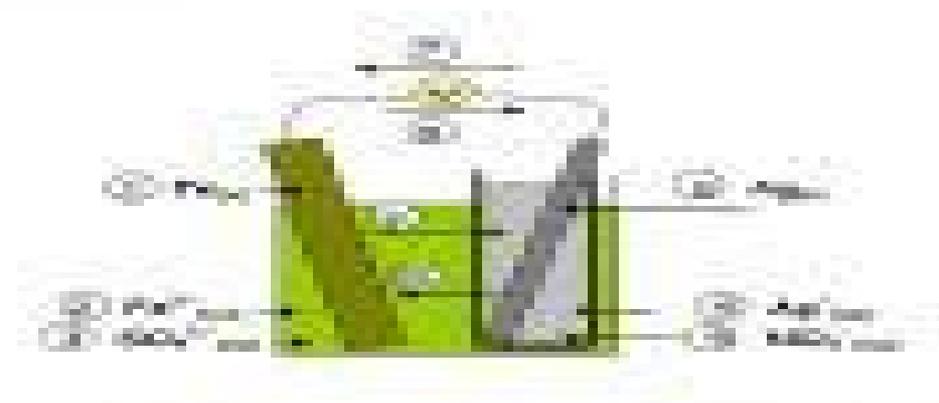


Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a. oxidant _____
- b. reductant _____
- c. oxidant _____
- d. reductant _____
- e. half-reaction (oxidation) _____
- f. half-reaction (reduction) _____

- g. write overall balanced redox reaction _____
- h. E_{cell} _____

i. Write overall cell process with phases. $\text{cell} = \text{anode} | \text{anode ion} || \text{cathode ion} | \text{cathode}$ (written as written in question with half-reactions for anode & cathode)

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Hussin A.Rothana



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