

CSEC Agricultural Science Double Award

Paper 2 – Structured Questions

Time: 2 hours 30 minutes

Total Marks: 100

Answer all questions in the spaces provided.

- (a) Define the term 'photosynthesis'. (2 marks)

(b) Describe two environmental factors that affect the rate of photosynthesis. (4 marks)

(c) Draw and label a diagram of a chloroplast. (4 marks)

(Total 10 marks)
- (a) List three functions of the root system in plants. (3 marks)

(b) Explain the process of transpiration and its importance to plants. (5 marks)

(c) Suggest two ways to reduce transpiration in crops. (2 marks)

(Total 10 marks)
- (a) What is soil erosion? (2 marks)

(b) Describe two methods farmers can use to control soil erosion. (4 marks)

(c) Explain one effect of soil erosion on agriculture. (4 marks)

(Total 10 marks)
- (a) Name two internal parasites that affect livestock. (2 marks)

(b) Explain two effects of internal parasites on livestock production. (4 marks)

(c) Outline two methods of controlling internal parasites. (4 marks)

(Total 10 marks)
- (a) Define crop rotation. (2 marks)

(b) List three advantages of practicing crop rotation. (3 marks)

(c) Design a four-year crop rotation plan including legumes and root crops. (5 marks)

(Total 10 marks)

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web the six $\text{NO}_3^- \text{aq}$ ions and the six $\text{Na}^+ \text{aq}$ ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{Ba}^{2+} \text{aq} + 2\text{PO}_4^{3-} \text{aq} \rightarrow \text{Ba}_3\text{P}_2\text{O}_7 \text{s}$

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web these spectator ions ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{h}^+ \text{aq} + \text{oh}^- \text{aq} \rightarrow \text{h}_2\text{O} \text{l}$ the spectator ions in $\text{koh} \text{aq}$ and $\text{hno}_3 \text{aq}$ are $\text{k}^+ \text{aq}$ and $\text{no}_3^- \text{aq}$ how to identify and list the

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web the $\text{ag}^+ \text{aq}$ and $\text{cl}^- \text{aq}$ ions become $\text{agcl} \text{s}$ but the $\text{na}^+ \text{aq}$ ions and the $\text{no}_3^- \text{aq}$ ions stay as $\text{na}^+ \text{aq}$ ions and $\text{no}_3^- \text{aq}$ ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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