

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe the components and direction of electron flow.

- a. anode _____
- b. cathode _____
- c. anion _____
- d. cation _____
- e. direction of electron flow _____
- f. direction of ion movement _____
- g. half-cell reduction equation _____
- h. cell _____
- i. Write overall cell equation with phases. $\text{Zn(s)} + \text{Cu}^{2+}(\text{aq}) \rightarrow \text{Zn}^{2+}(\text{aq}) + \text{Cu(s)}$ (do not use arrows, include all coefficients for balanced equation, correct 25°C notation, and phases, include state abbreviations for liquid & gaseous)

Voltaic Cells Study Guide

RS Peters



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