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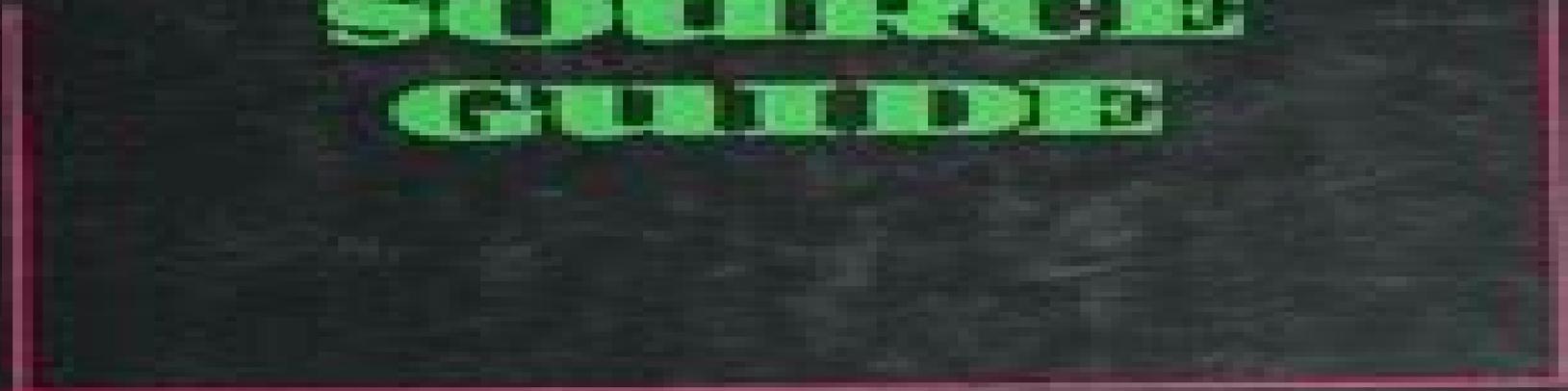


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Health A Multimedia Source Guide

G. Kim Dority



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web the pcl₃ lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is
from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note
the lone pair dots without bonds on top of p just like for n in the previous example for nh₃ chlorine cl

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