

**Unit 1 Study Guide — ANSWER KEY**

1. Define scientific method: **a procedure for collecting, testing, and communicating information**
2. Define hypothesis: **a suggested answer to the question, an educated guess**
3. Define experiment: **a scientific test so see what will happen**
4. Define dependent variable: **the variable that you do NOT control. (on a graph is usually on the Y axis)**
5. Define independent variable: **the variable that I control. (on a graph is usually on the X axis)**
6. Define control: **the part of the experiment where everything is left the same.**
7. Define observation: **information collected with any of your 5 senses**
8. Define density: **the ratio of mass per unit volume (D = m/v)**
9. Define accuracy: **the measure of how close your data is to the ACTUAL or true value.**
10. Define precision: **a measure of how close your data (measurements) are to each other.**
11. Why do scientists use SI units of measurement? **So that measurements can be understood internationally (around the world).**

12. List (in the correct order) the steps of the scientific method.

1. make an observation	4. test hypothesis (experiment)
2. ask a question	5. analyze data/form a conclusion
3. form a hypothesis	6. develop a theory

13. What mnemonic device is used to remember the metric system?

**K H Da B d c m**  
**g, m, L**

14. What is the sentence (mnemonic) that will help you remember the METRIC prefixes in order?

**King Henry Died By Drinking Chocolate Milk**

15. Fill in the table below describing the metric system:

	<b>Kilo</b>	<b>Hecto</b>	<b>Deca</b>	<b>Base</b>	<b>Deci</b>	<b>Centi</b>	<b>Milli</b>
Abbreviation	<b>K</b>	<b>H</b>	<b>Da</b>	<b>g, m, L</b>	<b>d</b>	<b>c</b>	<b>m</b>
Meaning in powers of 10	<b><math>10^3</math></b>	<b><math>10^2</math></b>	<b><math>10^1</math></b>	<b><math>10^0</math></b>	<b><math>10^{-1}</math></b>	<b><math>10^{-2}</math></b>	<b><math>10^{-3}</math></b>

Convert the following:

- |                                 |                                    |
|---------------------------------|------------------------------------|
| 16. 25L to <u>25000</u> mL      | 21. 25 cm to <u>0.25</u> m         |
| 17. 2.3 km to <u>230,000</u> cm | 22. 56784 mm to <u>0.056784</u> km |
| 18. 0.045 g to <u>45</u> mg     | 23. 15 g to <u>0.015</u> kg        |
| 19. 0.35 kg to <u>350</u> g     | 24. 3.452 kL to <u>3452</u> L      |
| 20. 300 mL to <u>0.300</u> L    | 25. 47 m to <u>47000</u> mm        |

**Scientific Notation:**

Convert the following to scientific notation:

- |            |                                      |            |  |
|------------|--------------------------------------|------------|--|
| 26) 0.005  | <u><math>5 \times 10^{-3}</math></u> | 30) 0.025  | <u><math>2.5 \times 10^{-2}</math></u> |
| 27) 5.050  | <u><math>5.05 \times 10^0</math></u> | 31) 0.0025 | <u><math>2.5 \times 10^{-3}</math></u> |
| 28) 0.0008 | <u><math>8 \times 10^{-4}</math></u> | 32) 500    | <u><math>5 \times 10^2</math></u>      |
| 29) 1,000  | <u><math>1 \times 10^3</math></u>    | 33) 5,000  | <u><math>5 \times 10^3</math></u>      |

# 700 Answer Key Study Guide

**Yi-Tong Ma**



**700 Answer Key Study Guide:**

## The Enigmatic Realm of **700 Answer Key Study Guide**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **700 Answer Key Study Guide** a literary masterpiece penned with a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those who partake in its reading experience.

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